# **Elephant Toothpaste**

#### Instructions

### Materials:

30% hydrogen peroxide, H<sub>2</sub>O<sub>2</sub> Potassium iodide, KI Dishwashing liquid Graduated cylinder Wood splint Match

### Safety Precautions:

30% hydrogen peroxide should be stored in the refrigerator when not in use. It is a strong oxidizer and very corrosive to the skin, eyes, and respiratory tract. Wear safety goggles and if possible perform this demonstration under a vented hood.

### Set Up:

- 1. This reaction produces a tremendous amount of foam so place the graduated cylinder inside a large tray to contain the waste.
- 2. Set out the reagents so that all the items are easily within your reach.

# Demo Procedure:

- 1. Carefully pour 50 mL of 30% H<sub>2</sub>O<sub>2</sub> into a tall graduated cylinder.
- 2. Add couple squirts of dishwashing detergent.
- 3. Add a small amount of KI to the cylinder and step away quickly.
- 4. The reaction is immediate; heat, steam and foam are produced and shoot out of the cylinder and spills over the side.
- 5. The foam is generated by the production of oxygen gas. This can be shown by using a glowing splint test.
- 6. Away from the foam, light a wood splint with a burning match. Blow out the flame but make sure that there is a visible glow on the tip of the splint.
- 7. Insert the glowing tip into the foam and watch as the splint re-ignite due to the concentrated oxygen gas in the foam. This can be repeated a few times using the same glowing splint.

<u>Waste Disposal</u>: Carry the entire tray with the cylinder and the foam to a deep sink. Allow the water to rinse away all the foam and the left over solution.