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ESTERS RCO₂R¹ VIII.H

Esters are named as 2 words: Alkyl...oate. The procedure for naming them is shown below.

Proc	edure for Naming Esters R-C-OR ¹
1.	Name the alkyl group \mathbb{R}^1 as the first word. The carbon bonded to the oxygen is C-1.
2.	Name the RC=O unit like a carboxylic acid, changing the ic acid ending to -oate.
3.	Name the ester as alkyl (word 1)oate (word 2).

Examples

1. Give the IUPAC name for **OBSERVATIONS**

CH₃CH₂CO₂CH₂CH₂CH(CH₃)₂

Redraw the structure $CH_3CH_2CO_2CH_2CH_2CH(CH_3)_2$

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Parent Group and Site	Ester at C-1	oate	
Longest Carbon Chain/Ring	Acyl chain, RC=O 3 carbons O-alkyl chain 4 carbons	propoate butyl	
# C=C or C≡C bonds and Site	None on either chain	butyl propanoate	
Final Word		propanoate	
Substituents and Sites	None on acyl chain Methyl at C-3 on O-alkyl	3-methylbutyl	
Alphabetizing substituents		3-methylbutyl	

SOLUTION Compound is 3-methylbutyl propanoate

CH₃CH=CHCO₂CH₂CH₃ 2. Give the IUPAC name for **OBSERVATIONS** CH₃CH=CHCO₂CH₂CH₃ Parent Functional Group(s): Ester → -oate Chain attached to C-O: CH₃CH₂- \rightarrow ethyl Carboxylate component (R-C=O): CH₃CH=CHCO₂ 2-butenoate \Rightarrow

SOLUTION Compound is ethyl 2-butenoate ed Material

$$C-C=C-C-C-C-C$$
4 3 2 1 1' 2'